

STEREOSELECTIVE SYNTHESIS OF THE FUNCTIONALIZED syn-1,3,5,7-TETRAOL

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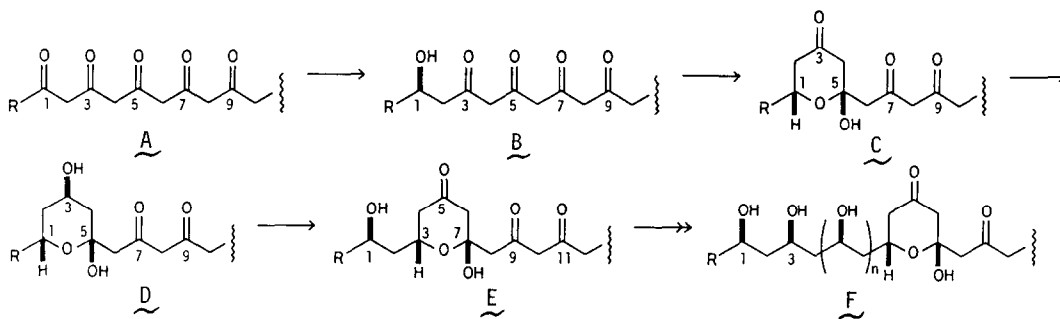
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Summary The syn-1,3,5,7-tetraol derivative 21 was synthesized with complete stereo-selection based on the stereoselective reduction of a six-membered β -keto hemiacetal and a subsequent hemiacetal ring opening

We previously reported a highly stereocontrolled reduction of the various acyclic ketones by means of $Zn(BH_4)_2$.¹ The stereoselectivity of the reduction is considered to be governed by the stability of a zinc mediated cyclic transition state. Therefore, if a functionalized acyclic ketone which is in equilibrium with a corresponding cyclic ketone is designed and the equilibrium lies to the latter, a well established stereocontrol in a cyclic system would be operative in this compound and after reduction, the resulting cyclic alcohol is convertible to the desired acyclic alcohol by simply shifting the equilibrium to this direction. We now report the synthesis of syn-1,3,5,7-tetraol with virtually complete stereoselection based primarily on this line.

A 1,3-polyol system is often found in polyoxygenated natural products such as polyene macrolide antibiotics and although several excellent methods have been reported,² the synthesis of stereochemically defined polyols is still highly required. These polyols are presumed to be produced biogenetically from the corresponding polyketide precursor A by NAD(P)H reduction. Our synthetic strategy is basically related to this biogenetic pathway and is schematically shown below. If a terminal ketone in A is reduced to an alcohol, the resulting B will be

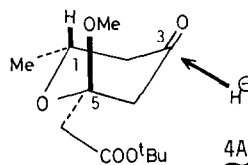


cyclized to the six-membered β -keto hemiacetal C and therefore, the reducing reagent may attack from the less hindered α -side producing the 3 β -hydroxy compound D. Opening of the hemiacetal ring in D and the successive ring closure between the newly produced C-3 hydroxyl group and C-7 ketone is now possible to produce the isomeric hemiacetal E. Since E contains again the same β -keto hemiacetal structure as C, the higher homologue of syn-1,3-polyol such as F will be

produced by repetition of the same reaction sequence. In this scheme, it is noteworthy that a six-membered β -keto hemiacetal is serving as a chiral syn-1,3-polyol chain producing unit.

In the actual synthesis, the above scheme was modified to a more practical one. The β -thioacetal- δ -lactone 1 [mp 78-80°C, NMR δ 1.41 (d, $J=6.4$ Hz, Me), 2.97 (d, $J=17.6$ Hz, C-4 α H)],³ prepared from ethyl β -hydroxy butyrate,⁴ was treated with $\text{CH}_3\text{COOt-Bu}$ and LDA in THF at -78°C to produce the hemiacetal 2 [mp 107-109°C, NMR δ 2.49, 2.50 (ABq, $J=15$ Hz, C-6 H_2)] as a single product in 94% yield.⁵ The hemiacetal 2 was converted to the acetal 3 by methylation with $\text{CH}(\text{OMe})_3$ and CSA, which was treated with NBS to afford the ketone 4 [NMR δ 3.25 (s, OMe)] in 74% yield. Stereostructure of 4 was determined based on the NOE measurement, enhancement of C-1 H signal (6.9%) observed upon irradiation of C-5 OMe suggested that C-1 H and C-5 OMe should be syn-diaxial.⁶ Therefore, in the reduction of 4, the hydride would attack the C-3 keto group preferentially from the less hindered equatorial side (see conformer 4A).

In order to enhance the stereoselectivity of the reduction, we chose the sterically bulky reagent "Selectride" as reducing agent.⁷ Reduction of 4 with K-Selectride in THF at -78°C gave only the expected 3 β -hydroxy isomer 5 [NMR δ 1.84 (dd, $J=14.4, 3.6$ Hz, C-4 α H)] in 96% yield. The configuration of the C-3 hydroxyl group was assigned as 3 β -axial by the coupling constant ($J=3.6$ Hz) between C-3 H and C-4 α axial H in the NMR spectrum. Subsequent treatment of 5 with excess 1,3-propanedithiol and $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in CH_2Cl_2 at -40°C produced the δ -lactone 6 [NMR δ 1.27 (d, $J=6.1$ Hz, Me), 2.93 (d, $J=17.1$ Hz, C-6 α H)] in 80% yield.⁸ In this reaction, transthioacetalization and lactonization took place successively releasing the C-1 hydroxyl group from the hemiacetal ring as expected. Since the product 6 has the same β -thioacetal- δ -lactone moiety as the starting material 1, the same reaction sequence described above is expected to be repeated.

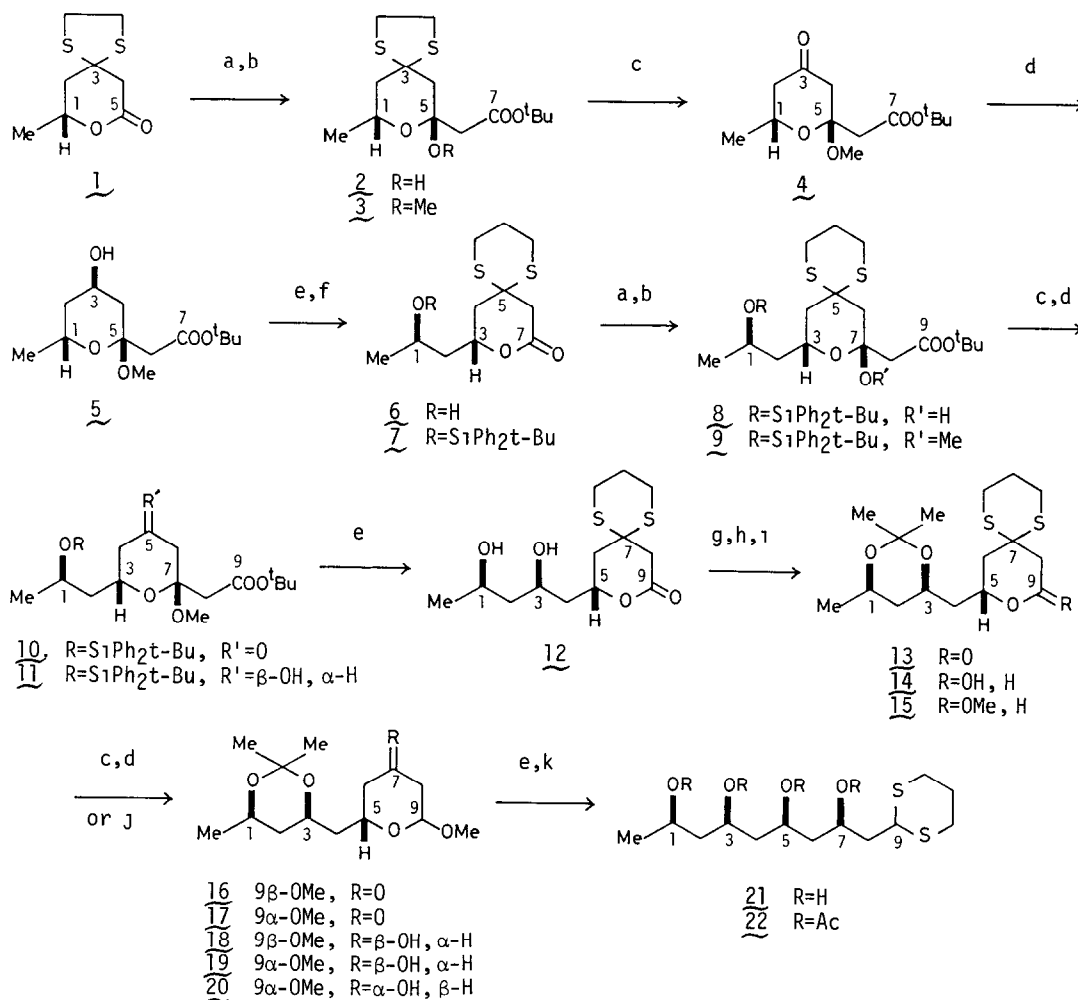


Indeed, the δ -lactone 6, after protection of the C-1 hydroxyl group as the *t*-butyldiphenylsilyl ether (81%), was successfully converted to the δ -lactone 12 in five steps (1~v), (i) introduction of C-2 unit ($\text{CH}_3\text{COOt-Bu/LDA/THF/-78}^\circ\text{C}$ 7→8, 74% yield), (ii) methylation of hemiacetal ($\text{CH}(\text{OMe})_3/\text{CSA}/\text{CH}_2\text{Cl}_2/\text{MeOH}/\text{rt}$ 8→9), (iii) dethioacetalization ($\text{NBS}/\text{AgNO}_3/\text{Na}_2\text{CO}_3/\text{aq MeCN}/0^\circ\text{C}$ 9→10), (iv) reduction ($\text{K-Selectride}/\text{THF}/-78^\circ\text{C}$ 10→11, 65% yield from 8), (v) transthioacetalization and successive lactonization ($1,3\text{-propanedithiol}/\text{BF}_3 \cdot \text{Et}_2\text{O}/\text{CH}_2\text{Cl}_2/-40$ ~ -20°C 11→12, 74% yield).^{8,9} K-Selectride reduction of 10 proceeded with complete stereoselection and afforded the desired 5 β -hydroxy isomer 11. The stereochemistry of the C-5 β hydroxyl group in 11 was confirmed based on the NMR data ($J_{5,6\alpha}=3.7$ Hz). The lactone 12 contains again the same β -thioacetal- δ -lactone moiety as 1 and 6, which shows that from this unique unit the ethanol moiety whose secondary hydroxyl group is syn to the β -hydroxyl group is newly produced through a series of the reactions (one cycle, five steps, 1~v). Therefore, further repetition of the same reaction sequence would, in principle, produce the higher homologue of syn-1,3-polyol.

Having confirmed the repeatability of the sequence (1~v), we then examined the termination of the reaction. After protection of the C-1,3 dihydroxyl group in 12 as an acetonide, 13 was reduced with DIBALH to give the lactol 14 in 60% yield. Subsequent methylation [$\text{CH}(\text{OMe})_3/\text{PPTS}$] of 14 followed by NBS treatment yielded a 1:1 mixture of the 9 β - and 9 α -methoxy ketones, 16 [NMR δ 3.37 (s, OMe)] and 17 [NMR δ 3.54 (s, OMe)], in 90% yield. The mixture of 16 and 17, without separation,¹⁰ was reduced with K-Selectride to give a mixture of 18, 19, and 20 in a

ratio of 100 : 84 : 32 (73% yield).¹¹ Namely, reduction of the 9 β -methoxy ketone 16 produced only the 7 β -hydroxy compound 18 as expected,¹² while the 9 α -methoxy ketone 17 gave the undesired 7 α -hydroxy isomer 20 along with the 7 β -hydroxy compound 19. On the other hand, reduction of the mixture of 16 and 17 with KS-Selectride in THF gave only the 7 β -hydroxy compounds 18 and 19 in a ratio of 1 : 1 (39% yield).¹³ Finally, on treatment with 1,3-propanedithiol and BF₃·Et₂O at 0°C, the 7 β -hydroxy compounds, 18 and 19, were effectively converted to the *syn*-1,3,5,7-tetraol 21 in 69% yield, which was acetylated (Ac₂O/py/rt) to give the *syn*-1,3,5,7-tetraacetate 22 [NMR δ 2.032, 2.057, 2.065, 2.068 (each s, 4xAc)]

A practical method for the synthesis of the functionalized *syn*-1,3-polyol system was thus established. The synthesis of optically active natural products containing *syn*-1,3-polyol system is now in progress.

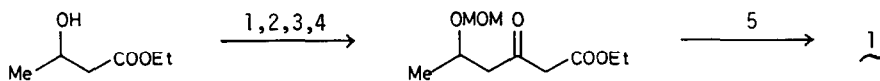


a CH₃COOt-Bu/LDA/THF/-78°C, **b** CH(OMe)₃/CSA/CH₂Cl₂/MeOH/rt, **c** NBS/AgNO₃/Na₂CO₃/aq MeCN/0°C, **d** K-Selectride/THF/-78°C, **e** 1,3-propanedithiol/BF₃·Et₂O/CH₂Cl₂, **f** t-BuPh₂S₁Cl/imidazole/DMF/rt, **g** Me₂C(OMe)₂/CSA/CH₂Cl₂/rt, **h** DIBAH/toluene/-78°C, **i** CH(OMe)₃/PPTS/CH₂Cl₂/rt, **j** KS-Selectride/THF/-30~-10°C, **k** Ac₂O/pyridine/rt

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References and Notes

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2. P. A. Bartlett and K. K. Jernstedt, J. Am. Chem. Soc., 99, 4829 (1977), P. A. Bartlett and K. K. Jernstedt, Tetrahedron Lett., 21, 1607 (1980), P. A. Bartlett, J. D. Meadows, E. G. Brown, A. Morimoto, and K. K. Jernstedt, J. Org. Chem., 47, 4013 (1982), G. Cardillo, M. Orena, G. Porzi, and S. Sandri, J. Chem. Soc. Chem. Comm., 465 (1981), M. Hirama and M. Uei, Tetrahedron Lett., 23, 5307 (1982), D. M. Floyd and A. W. Fritz, ibid., 22, 2847 (1981), J. M. Finan and Y. Kishi, ibid., 23, 2719 (1982), P. Ma, V. S. Martin, S. Masamune, K. B. Sharpless, and S. M. Viti, J. Org. Chem., 47, 1378 (1982), K. C. Nicolaou and J. Uenishi, J. Chem. Soc. Chem. Comm., 1292 (1982), K. Narasaka and H. C. Pai, Chem. Lett., 1415 (1980)
3. The NMR spectra of the compounds (1~22) were obtained in CDCl_3 on a JEOL FX-400 or GX-400 spectrometer (400 MHz).
4. The starting material 1 was synthesized by the route shown below. The optically active β -hydroxy esters are readily prepared. Therefore, when these compounds are used, the optically active 1 and thence optically active 1,3-polyols can be synthesized.



- 1) $\text{MeOCH}_2\text{Cl}/1\text{-Pr}_2\text{NEt}$, 2) DIBAH (or LiAlH_4 , PCC), 3) $\text{CH}_3\text{COOt-Bu/LDA}$,
4) Jones oxidation, 5) ethanedithiol/ $\text{BF}_3 \cdot \text{Et}_2\text{O}$

5. cf. A. J. Duggan, M. A. Adams, P. J. Brynes, and J. Meinwald, Tetrahedron Lett., 4323 (1978)
 6. The present result was confirmed by NOE difference spectrum
 7. cf. H. C. Brown and S. Krishnamurthy, Aldrichimica Acta, 12, No 1, 3 (1979)
 8. The reaction mixture was directly separated by silicagel column or PTLC after excess 1,3-propanedithiol was removed in vacuo
 9. The reaction was carried out until the silyl group was completely removed (ca. 7 hr).
 10. No attempt was made to separate the mixture, although the analytical TLC indicated the two separated spots
 11. The ratio was determined by NMR data. The three isomers, 18, 19, and 20, are separable on preparative TLC. NMR 18 δ 3.390 (s, OMe), 4.07 (m, $W_{h/2}=10$ Hz, C-7 H), 19 δ 3.493 (s, OMe), 4.33 (m, $W_{h/2}=7.4$ Hz, C-7 H), 20 δ 3.489 (s, OMe), 3.84 (m, $W_{h/2}=24$ Hz, C-7 H)
 12. cf. S. Danishefsky, J. F. Kerwin Jr., and S. Kobayashi, J. Am. Chem. Soc., 104, 358 (1982)
 13. Although TLC analysis of the reaction mixture showed almost no by-product, the yield was rather low. The yield is not optimized yet.
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